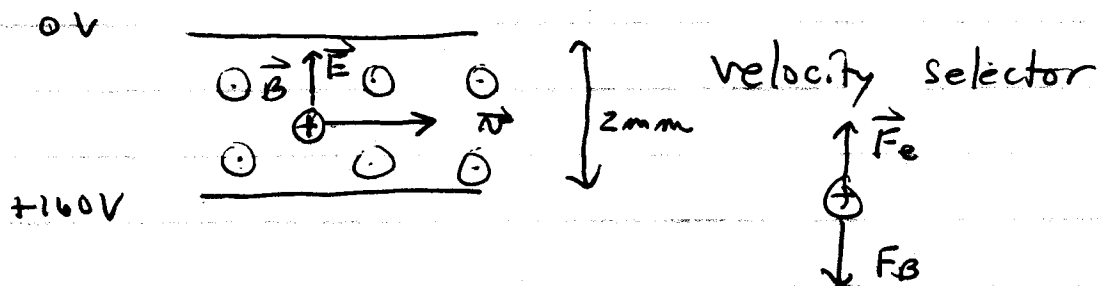


48. Before entering a mass spectrometer, ions pass through a velocity selector consisting of parallel plates separated by 2.0 mm and having a potential difference of 160 V. The magnetic field between the plates is 0.42 T. The magnetic field in the mass spectrometer is 1.2 T. Find (a) the speed of the ions entering the mass spectrometer and (b) the difference in the diameters of the orbits of singly ionized ^{238}U and ^{235}U . (The mass of a ^{235}U ion is 3.903×10^{-25} kg.)

#48

Mass Spectrometer



$$\text{Want } F_e = F_b$$

$$eE = evB$$

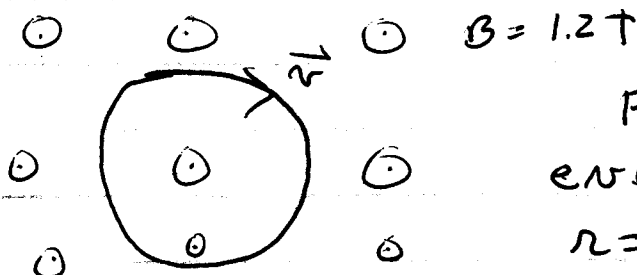
$$v = E/B$$

$$\text{What is } E? \quad E = \frac{\Delta V}{d} = \frac{160\text{V}}{0.002\text{m}} = 8.0 \times 10^4 \text{V/m}$$

$$B = 0.42\text{T (given)}$$

$$v = E/B = 1.905 \times 10^5 \text{m/s}$$

Next: what is radius in spectrometer?



$$F = ma$$

$$evB = m v^2 / r$$

$$r = \frac{mv}{eB} \Rightarrow r =$$

$$m_{235} = 3.903 \times 10^{-25} \text{kg (given)}$$

$$m_{238} = \frac{238}{235} m_{235}$$

$$r_{235} = \frac{(3.903 \times 10^{-25} \text{kg})(1.905 \times 10^5 \text{m/s})}{(1.602 \times 10^{-19} \text{C})(1.2 \text{T})} = 0.387 \text{m}$$

$$r_{238} = \frac{238}{235} r_{235} = 0.392 \text{m}$$

$$\Delta r = 0.0049 \text{m} = 4.9 \text{mm}$$